2005 Vol. 7, No. 22 4859–4861

## Preparation and Reaction of 2-Aryl-3-silyl-1,3-butadiene

Zenichi Ikeda, Koichiro Oshima, and Seijiro Matsubara\*

Department of Material Chemistry, Graduate School of Engineering, Kyoto University, Kyoutodaigaku-katsura, Nishikyo, Kyoto 615-8510, Japan matsubar@orgrxn.mbox.media.kyoto-u.ac.jp

Received July 2, 2005

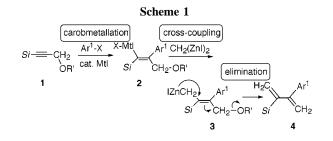
## **ABSTRACT**

$$H_2C(ZnI)_2$$
,  $Ar^1$ -I  $Ar^2$ 

A conjugated  $\pi$ -electron compound, 2-aryl-3-silyl-1,3-butadiene, was easily prepared from 1-benzyloxy-3-silyl-2-propyne, bis(iodozincio)methane, and an aryl halide in the presence of nickel catalyst. A subsequent cross-coupling reaction of the product with another aryl halide gave an unsymmetrical 2,3-diaryl-1,3-butadiene efficiently.

Conjugated  $\pi$ -electron compounds such as aryl-substituted polyenes attract a lot of attention from material scientists.<sup>1</sup> The most common and effective strategy to construct the aryl-substituted polyene skeleton has been stepwise assembling of alkene and arene units by means of crosscoupling reaction.<sup>2</sup> One drawback of the methodology is that some undesirable  $\pi$ -bond isomerization, including polymerization, may occur under the cross-coupling reaction conditions. To prevent such isomerization, researchers have tuned and modifed the organometallic compound that is one of the cross-coupling partners. For example, Denmark reported recently an efficient coupling of unsymmetrical 1,4bissilyl-1,3-butadiene with an aryl halide to construct 1-silyl-4-aryl-1,3-butadiene. The compound can be transformed into an unsymmetrical 1,4-diaryl-1,3-butadiene or the conjugated polyenes through further cross-coupling reaction.<sup>3</sup> Hiyama showed an efficient preparation of 2,3-bisboryl-1,3-butadiene, which is a potent precursor for 2,3-diaryl-1,3-butadiene or dendralene. 4,5 In these methods, bissilvlated or -borated 1,3butadiene is the key intermediate, on which selective crosscoupling in the right sequence was performed. Several examples that are based on carbometalation have also been

reported.<sup>2b,6</sup> In these methods, an intermediate that is formed by transition metal catalyzed carbometalation between alkyne and aryl halide undergoes an additional cross-coupling reaction or Heck reaction. We have reported a series of synthetic studies of bis(iodozincio)methane.<sup>7</sup> This *gem*-dizinc reagent is prepared from diiodomethane, zinc, and a catalytic amount of lead(II) chloride as reported previously.<sup>8</sup> If an intermediate **2** via a carbometalation of aryl halide to



propargyl derivative 1 (Scheme 1) undergoes a cross-coupling reaction with bis(iodozincio)methane, the obtained

<sup>(1)</sup> Conjugated Polymers and Related Materials; Salaneck, W. R., Lundström, I., Rånby, B., Eds.; Oxford University Press: Oxford, 1993.

<sup>(2) (</sup>a) Babudri, F.; Farinola, G. M.; Fiandanese, V.; Mazzone, L.; Naso, F. *Tetrahedron* **1998**, *54*, 1085–1094. (b) Shibata, K.; Satoh, T.; Miura, M. *Org. Lett.* **2005**, *7*, 1781. (c) Itami K, Mineno M, Muraoka N, Yoshida J. *J. Am. Chem. Soc.* **2004**, *126*, 11778.

<sup>(3)</sup> Denmark, D. E.; Tymonko, S. A. J. Am. Chem. Soc. 2005, 127, 8004. (4) (a) Shimizu, M.; Tanaka, K.; Kutrahashi, T. Shimono, K.; Hiyama, T. Chem. Lett. 2004, 33, 1066. (b) Shimizu, M.; Kurahashi, T.; Hiyama, T. Synlett 2001, 1006.

<sup>(5)</sup> Hopf, H. Angew. Chem., Int. Ed. 2001, 40, 705.

<sup>(6) (</sup>a) Zhou, C.; Emrich, D. E.; Larock, R. C. *Org. Lett.* **2003**, *5*, 1579. (b) Zhang, X.; Larock, R. C. *Org. Lett.* **2003**, *5*, 2993. (c) Thadani, A. N.; Rawal, V. H. *Org. Lett.* **2002**, *4*, 4317. (d) Pal, M.; Parasuraman, K.; Subramanian, V.; Dakarapu, R.; Yeleswarapu, K. R. *Tetrahedron Lett.* **2004**, *45*, 2305. (e) Pottier, L. R.; Peyrat, J.-F.; Alami, M.; Brion, J.-D. *Synlett* **2004**, 1503.

<sup>(7) (</sup>a) Matsubara, S.; Oshima, K. *Proc. Jpn. Acad.* **2003**, *79*, 71. (b) Matsubara, S.; Oshima, K. In *Modern Carbonyl Olefination*; Takeda, T., Ed.; Wiley-VCH: Weinheim, 2004; p 200. (c) Hirayama, T.; Oshima, K.; Matsubara, S. *Angew. Chem., Int. Ed.* **2005**, *44*, 3293.

allylzinc derivative **3** will be transformed into a 1,3-butadiene derivative **4** via elimination of zinc alkoxide. Accoding to this method, 2-aryl-3-silyl-1,3-butadiene, which is a convenient precursor for unsymmetrical 2,3-diaryl-1,3-butadiene, can be prepared in one step. Treatment of 3-trimethylsilyl-1-bezyloxy-2-propyne (**1a**) with bis(iodozincio)methane and iodobenzene in the presence of some palladium catalyst that was prepared from Pd<sub>2</sub>dba<sub>3</sub>·CHCl<sub>3</sub> and PAr<sub>3</sub> (Ar = phenyl, *p*-tolyl, 2-furyl, 3,5-bis(trifluoromethyl)phenyl) resulted in quantitative recovery of alkyne **1a**.<sup>9</sup> As shown in Scheme 2,

however, use of Ni(II) catalyst gave the desired 2-aryl-3-silyl-1,3-butadiene **6a**. Optimization of the yield of **6a** by tuning the nickel(II) catalyst was examined, as shown in Scheme 2. The catalyst prepared from 2 mol % of Ni(cod)<sub>2</sub> and 3 mol % of P(2-furyl)<sub>3</sub> improved the yield of **6a** to 99%.

Using the nickel catalyst, we examined the reaction of **1a** with bis(iodozincio)methane (**5**) and various aryl iodides, as shown in Scheme 3. It is notable that the reaction of 4-bromo-

1-iodobenzene gave the 4-bromophenyl substituted butadiene **6e** in excellent yield. The product may be a substrate for another cross-coupling reaction as aryl bromide.

The trimethylsilyl group is not an attractive group as an organometallic reagent for the Hiyama coupling reaction. The various organosilyl compounds, instead of the trimethylsilyl group, have often been used in attempts to form a hyperco-

ordinated silyl group, which is the key intermediate for the cross-coupling. <sup>10</sup> Trost demonstrated that the benzyldimethylsilyl group benefits the cross-coupling reaction, as it is easily transformed into the fluorodimethylsilyl group, which is the important hypercoordinated silyl group precursor. <sup>11</sup> We had also reported that the (2,6-dimethoxyphenyl)-dimethylsilyl group can be transformed into the fluorodimethylsilyl group with BF<sub>3</sub>—acetic acid complex. <sup>12</sup> As shown in Scheme 4, the propyne derivatives with these silyl groups

## Scheme 4

$$\begin{array}{c|c} -\dot{S}i & \longrightarrow & \begin{array}{c} H_2C(ZnI)_2\ (5), Ph\text{-I} \\ (1.2\ eq) & (1.2\ eq) \end{array} \\ OCH_2Ph & \begin{array}{c} \text{Ni Catalyst} \\ \text{THF, 25 °C, 18 h} & -Si\text{--}7 \\ \end{array} \\ \text{Ni Catalyst: Ni(cod)}_2\ (2.0\ mol\ \%) + P\left( \begin{array}{c} O\\ O \end{array} \right)_3\ (3.0\ mol\ \%) \end{array}$$

Si:	yield of 7
PhMe <sub>2</sub> Si ( <b>1b</b> )	76% ( <b>7b</b> )
(PhCH <sub>2</sub> )Me <sub>2</sub> Si ( <b>1c</b> )	97% ( <b>7c</b> )
(2,6-(MeO) <sub>2</sub> C <sub>6</sub> H <sub>3</sub> )Me <sub>2</sub> Si ( <b>1d</b> )	92% ( <b>7d</b> )

(1b-d) were examined for the reaction. In each case, the 2-phenyl-3-silyl-1,3-butadiene 7b-c was obtained in good yield.<sup>13</sup>

As shown in Scheme 5, 2,3-diaryl-1,3-butadiene was prepared using **7d**. A transformation of **7d** into fluorosilane derivative **8**,<sup>12</sup> followed by cross-coupling reaction with 4-iodotoluene, gave the 2,3-diaryl-1,3-butadiene **9**.

This sequential carbometalation and cross-coupling method can be also applied to diiodoarene. The method will give an arene with two silylbutadienyl groups. Diiodo arenes 10a-c were treated with 1a and bis(iodozincio)methane (5) in the presence of nickel catalyst to give the highly conjugated  $\pi$ -electron compounds 11a-c in good yields (Scheme 6).

We propose in Scheme 1 that a mechanism of these transformations may be initiated by carbonickelation of arylnickel iodide, which is formed by oxidative insertion of low-valent nickel to aryl iodide. Two other routes are possible for the formation of 2-aryl-3-silyl-1,3-butadiene. One con-

4860 Org. Lett., Vol. 7, No. 22, 2005

<sup>(8)</sup> Matsubara, S.; Oshima, K.; Matsuoka, H.; Matsumoto, K.; Ishikawa, K.; Matsubara, E. *Chem. Lett.* **2005**, *34*, 952.

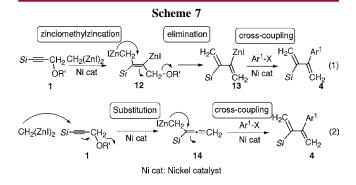
<sup>(9)</sup> Matsubara, S.; Ukai, K.; Toda, N.; Utimoto, K.; Oshima, K. Synlett **2000**, 995.

<sup>(10)</sup> Hiyama, T.; Hatanaka, Y. Pure Appl. Chem. 1994, 66, 1471.

<sup>(11)</sup> Trost, B. M.; Machacek, M. R.; Ball, Z. Org. Lett. 2003, 5, 1895.
(12) Utimoto, K.; Otake, Y.; Yoshino, H.; Kuwahara, E.; Oshima, K.;
Matsubara, S. Bull. Chem. Soc. Jpn 2001, 74, 753.

<sup>(13)</sup> Takahashi, T.; Xi, Z.; Fischer, R.; Huo. S.; Xi, C.; Nakajima, K. J. Am. Chem. Soc. 1997, 119, 4561.

tains an iodozinciomethylzincation as the first step (eq 1 in Scheme 7), and the other is formation of allenylzinc  ${\bf 14}$  as an intermediate by nickel-catalyzed  $S_N2'$  reaction (eq 2 in Scheme 7). In some cases, nickel-catalyzed carbozincation is a reasonable route, as reported previously. A route similar to that in eq 2 was shown in our previous report about



the reaction between propargylic bromide and bis(iodozincio)methane (5) in the presence of palladium catalyst, which affords butadienyl zinc species in situ. In the present case, however, treatment of 1a with bis(iodozincio)methane (5) in the presence of nickel catalyst prepared from Ni(cod)<sub>2</sub> and (2-furyl)<sub>3</sub>P resulted in the quantitative recovery of 1a. For these reasons, we prefer Scheme 1. Even in this case, the carbometalation of arylnickel iodide to propargylic ether 1 occurs under equilibrium and transmetalation of the alkenyl nickel species 2 with bis(iodozincio)methane should be the crucial reaction for these transformations.<sup>17</sup>

**Acknowledgment.** This work was supported financially by the Japanese Ministry of Education, Science, Sports, and Culture. Financial support from Chugai Pharmaceutical Co., Ltd. and Takahashi Industrial and Economical Research Foundation are also acknowledged.

**Supporting Information Available:** Experimental details and spectral data. This material is available free of charge via the Internet at http://pubs.acs.org.

OL051557S

(17) Nakao, Y.; Oda, S.; Hiyama, T. J. Am. Chem. Soc. 2005, 127, 13904.

Org. Lett., Vol. 7, No. 22, 2005

<sup>(14)</sup> Negishi E.; Miller, J. A. J. Am. Chem. Soc. 1983, 105, 6761.

<sup>(15)</sup> Stüdemann, T.; Ibrahim-Ouali, M.; Knochel, P. Tetrahedron Lett. 1998 54 1299

<sup>(16)</sup> Kimura, M.; Ezoe, A.; Mori, M.; Tamaru, Y. J. Am. Chem. Soc. **2005**, 127, 201.